Ion beam preparation of Au nanoparticles in crystalline GaN and ZnO semiconductors

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Wurtzite crystalline semiconductors, such as gallium nitride (GaN) and zinc oxide (ZnO), are widely used in electronics, energy storage devices, sensors, and as photocatalysts due to their attractive optoelectronic properties^{1,2}. However, their wide bandgaps limit their functionality mainly to the UV region. Modification with metallic gold (Au) nanoparticles can enhance charge-carrier mobility and device performance³. Moreover, Au nanoparticles exhibit visible light absorption due to surface plasmon resonance (SPR). thereby extending the application range of these semiconductors⁴. Ion implantation is a promising direct method to synthesize metal nanoparticles within solids, offering high purity and stability⁵. However, this process may introduce structural damage and impact material properties, especially in crystals, necessitating a detailed investigation. In this work, Au nanoparticles were formed via 1.85 MeV and 1 MeV Au ion implantation in crystalline GaN and ZnO, respectively. Implantation in ZnO caused less lattice damage than in GaN, and the Au ions exhibited a Gaussian depth distribution as characterized by Rutherford Backscattering Spectroscopy in channeling mode (RBS-C). GaN showed a more complex damage profile, resulting in a multimodal Au depth distribution, also confirmed by RBS-C. Transmission electron microscopy revealed distinct stages of Au nanoparticle formation in GaN. These stages, along with crystal quality, significantly influenced the optical response, as characterized by absorption/reflection measurements and photoluminescence. In particular, implanted GaN exhibited new blue (2.8 eV) and red (2 eV) emission bands, attributed to the combined effects of SPR and implantation-induced defects.

1) J. L. Weyher et. al, *Applied Surface Science*, 2019, 466. https://doi.org/10.1016/j.apsusc.2018.10.076

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²⁾ N. Morales-Flores et. al, *Applied Catalysis A: General*, 2011, 394. https://doi.org/10.1016/j.apcata.2011.01.011

- 3) H. Zheng et. al, *Physica Status Solidi* (a), 2023, 220. https://doi.org/10.1002/pssa.202300037
- 4) S. Lu et. al, ACS Appl. Mater. Interfaces, 2014, 6. https://doi.org/10.1021/am503442c
- 5) M. C. Salvadori et. al, *Applied Surface Science*, 2014, 310. https://doi.org/10.1016/j.apsusc.2014.03.145